## A New Reaction, a One-step Synthesis of N-Substituted Thioamide from Nitriles

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We have found that the reaction of nitriles with N-substituted dithiocarbamates constitutes a facile one-step synthesis of N-substituted thioamides. Thus, when a mixture of benzonitrile, dimethylammonium N, N-dimethyldithiocarbamate, and benzene was heated for several hours under pressure by shaking, N, N-dimethyl benzthioamide was obtained; it proved to be identical with an authentic sample prepared by the thionation of N, N-dimethylbenzamide with phosphorus pentasulfide.<sup>12</sup>

This reaction can be successful with a number of aromatic nitriles and aralkyl nitriles, and with a variety of heterocyclic nitriles, as is shown in Table 1.

The reaction mechanism will be described and discussed in detail in the near future elsewhere.

**Synthesis of N, N-Dimethylthiobenzamide.** 5.15 g (0.05 mol) of benzonitrile was added to a suspension of 8.5 g (0.05 mol) of dimethylammonium N, N-dimethyldithiocarbamate in 25 ml of

absolute benzene. The mixture was heated in an autoclave by shaking it at  $150^{\circ}\text{C}$  for 3 hr. The reaction mixture was then allowed to stand overnight. To this 100 ml of ether were added in order to remove a small quantity of an insoluble material. The ethereal solution was dried over anhydrous sodium sulfate, the solvent was removed on a rotating evaporator, and the resulting red-brownish residue was fractionated under reduced pressure. N, N-Dimethylbenzthioamide distilled over at 153— $163^{\circ}\text{C}/7$ —8 mmHg as a light yellow oil which was soon solidified. After recrystallization from methyl acetate, light yellow crystals (mp 59— $64^{\circ}\text{C}$ ) resulted. UV:  $\lambda_{max}^{\text{CH}_3\text{OH}}$  278 m $\mu$  ( $\varepsilon_{max}$  9181), IR:  $1515 \text{ cm}^{-1}$ ,  $1295 \text{ cm}^{-1}$ ,  $1140 \text{ cm}^{-1}$  (-C-N $\langle$ )  $1450 \text{ cm}^{-1}$ 

cm<sup>-1</sup>, 1490 cm<sup>-1</sup> (benzene ring).  $R_f$  value (thin layer, silica gel) 0.916 (EtOH: CHCl<sub>3</sub>: AcOEt= 4:3:2).

TABLE 1

Nitrile	N-Dimethylcarbamate	Solvent	Reaction temp., °C	Reaction time, hr	Yield %
Benzo-	Dimethylamine salt	Benzene	150—160	3	79.9
Benzo-	Dimethylamine salt	Dimethyl- cellosolve	145—150	3	35.9
Benzo-	Dimethylamine salt	Pyridine	150-155	3	35.6
Phenylaceto-	Dimethylamine salt	Xylene	190-210	3	35.5
Butyro-	Dimethylamine salt	Xylene	190-210	3	48.6
3-Cyano-4- methoxymethyl- 6-methyl 2- pyridone	Dimethylamine salt	Xylene	150—170	3	9.3

<sup>1)</sup> K. Kindler, Ann., 431, 210 (1923).